

Burning Old Fuels in New Ways Recent Projects in Combustion Research

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Combustion Research at UCI





Burning Old Fuels in New Ways

- Coal combustion
 - control of carbon-in-ash from pulverized coal and coal/ biomass combustion
 - QCL absorption measurements for zonal control of combustion
 - control of mercury emissions
- Combustion in chambers
 - Improved miniature liquid fueled combustion systems
 - flame holding in a model turbine burner channel
- Electrical properties of flames
 - small diffusion flames
 - partially premixed flames
- Alternative fuels
 - biofuel combustion
 - methane and propane hydrate combustion

Fluid/Thermal Sciences

- Fuel preparation and sprays
 - controlled breakup of droplet streams
 - fuel injection for small-scale homogeneous charge compression ignition (HCCI) engine
- Multiphase flow and particle interaction
 - particle interaction in viscoelastic fluid
 - turbulence effects on droplet vaporization
- Biomedical applications
 - Dental laser drilling, sterilization, and therapy

Lasers, Flames & Aerosols Research Group

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Simplified Systems



Sprays: droplet stream combustion, supercritical behavior, emulsion flows;

Gas burners: flat flames, slot burners, electric flames; Coal combustion: entrained flow reactor, ammonia bisulfate condensation, mercury formation and capture; Aerosols: lon driven flows, indoor air modeling.

Goals



Simplified systems that mimic a specific behavior of interest of a more complex systems, such as spray combustion, utility coal plants, and IC engines, are studied to obtain fundamental information.

In many cases, numerical simulations are developed in conjunction with the experiments to provide insight into influencing factors. Although our motivation remains in better understanding of a practical systems, diagnostic tools utilized to meet these goals are concurrently being evaluated in to prove flexibility and accuracy in non-ideal settings.

Diagnostics in use



LDV, PIV, PCSV, PLIF, CARS, schlerien, high speed videography,

shadowgraphy, the deconvolution of mie scattered fringes, polarization sizing technique, and fluorescence techniques.

eneral Electric and Industry-University Cooperative Research Program has begun testing in the newly built 2 meter entrained flow reactor. This

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Ifa.eng.uci.edu

Non-fossil energy use grows rapidly, but fossil fuels still provide 79 percent of total energy use in 2030





EIA Annual Energy Outlook 2009 Reference Case Presentation -- December 17, 2008



Char Burnout Kinetics

Background

- A major waste product after the combustion of coal and biomass is ash.
- High carbon content (> 5%) prevents the sale of the fly ash to cement industries.

Objectives

- Predict relationships between boiler operating conditions and residual carbon in ash.
- Predict latter stages of carbon burnout.

Entrained Flow Reactor

- 6 ft. length and 8 in. dia. combustion zone.
- Coal feed rate is 6.3 g/min.
- Residence time is 2 secs at 1 m/s flow rate.

Numerical Modeling

- A numerical model of coal burning in EFR to evaluate common char burnout kinetic models, including CBK.
- The CFD simulation includes heat transfer, fluid dynamics, as well as coal/ash particle tracking and combustion chemistry.



Current Image of EFR

Solid Model of UCI EFR



Temperature Contour (K)

Particle traces colored by residence time (second)

Char Burnout Kinetics

Diagnostic Instruments

– Loss On Ignition Measurements

The percentage of carbon in ash can be found.

 $LOI = \frac{Carbon_in_dry_ash}{Carbon + Ash_residue}$

– Sample Extraction Probe

- Inserted through ports on side for extraction at nearly any residence time.
- Water cooled probe with Nitrogen gas quench freezes sample at extraction.

– TESTO Gas Analyzer

Oxygen profile (radial, axial)

– GE Oxytrak 390 Flue Gas Analyzer

Continuous measurement of Oxygen and Combustibles concentration at the exit.

– Scanning Electron Microscope

Coal morphology analysis and individual particle size measurement.



Comparison of Char Burnout Models

- The coal particle life histories were studied by applying 3 char burnout models (Fluent CBK, External CBK and Kinetic-Diffusion Limited Model) and timetemperature, stoichiometric history of particles were compared with one another.
- Although CBK is the most rigorous model it has not been validated.
- We have all different tools and numerical models to predict char burnout, but experimental results will give us some idea how accurate these tools are.



Tomographic analysis of quantum cascade laser absorption by carbon monoxide

 Spectral absorption- environmental monitoring, industrial process monitoring, medical diagnostics, defense and security



 Comparing numerical prediction of CO concentration in a coarse of line-ofsight values by Beer-Lambert Law.





$$A = \ln \left[\frac{I_o}{I}\right] = -\ln \left[\frac{I}{I_o}\right] = \varepsilon c l = \alpha l$$

- A absorbance
- I transmitted light
- lo incident light
- ε molar absorptivity
- L path length
- α absorption coefficient

Tomographic analysis of quantum cascade laser absorption by carbon monoxide



Mercury Emission Control

Motivation

- Anthropogenic mercury emission from power plants
 – approx. 48 tons per year
- USGS survey found mercury in every fish tested from 291 streams; fish at more than 25% of sites had mercury above EPA safe level
- A typical 250 MW coal-fired power requires more than \$1 million in commercial sorbent annually
- Effective & inexpensive technology for mercury capture is needed

Goals

Simultaneous CO2 sequestration and mercury capture

Mercury Removal from Flue Gas by Aqueous Precipitation

Background

- Natural occurrence of mercury Hg⁰, as vapor in atmosphere Hg₂²⁺, Hg²⁺, as part of inorganic and organic compounds in water, soil, plants and animals
- average Hg content in coal in US is 0.06 ppm
- Recent estimation of mercury emission from power plants in US, 40-52 tons/yr
- United Nations' estimation of mercury speciation from coal combustion plant in 1995

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Hg^050%insolubleHg^{2+}40%solubleHg_p10%
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- Carbonate Mineralization by Aqueous Precipitation (CMAP)
 - CO₂ sequestration (Calera corp.)
 - Seawater contains the following pertinent chemical species:

Ca²⁺, Mg²⁺, CO₃²⁻, HCO₃⁻, (CO₂)_{aq}, H₂CO₃, H⁺ and OH⁻

Precipitation occurs when [M²⁺][CO₃²⁻] > Ksp

System is running under strong alkaline conditions to ensure the equilibrium favors the formation of [CO₃²⁻]







Example Test Run – Mercury Capture



15% of elemental mercury was removed with very stable result.

30% of oxidized mercury seemed to have been removed, however, the reading at outlet dropped continuously. This might be a contamination issue, since mercury sticks to metal and even teflon tubing. After prolonged operation, oxidized mercury at outlet lowered to a background level at around 6ug/m3. It suggests that this process actually has the ability to remove a high ratio of Hg2+.

The CMAP process is comparable with other wet scrubbing processes, i.e. FGD, with respect to mercury removal.

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Miniature Liquid Fuel Film Combustor with Air injection



All dimensions in mm

Collaboration with Professor Sirignano

CARS Temperature Measurements above a Miniature Liquid Film Combustor



Liquid Film Combustor – with air injection





Typical swirl flow of a tubular flame





When fuel pool is present, flame condition oscillates between internal and external burning

Horizontal orientation has no effect

Swirl Vane Design For Miniature Fuel Film Combustor

Motivation:

Claudio Giani

- High specific power and energy density
- Portable power system applications with power range of 10 to 1000 W



Concept:

A fuel film along the chamber with swirling air to enhance mixing and recirculation. The model of combustion is based on triple flame theory and the tubular flame.

Objective:

- Design swirl vane by rapid prototyping
- Confine the flame inside the chamber
- Improve miniaturization

Results:



Burner components







Swirl vanes designed



Tubular flame top view



Stable flame

Claudio Giani

Turbine Burner Combustion

- Improve gas turbine engine performances (specific thrust, thrust/ power to volume ratios, SCF, etc...) by adding energy in the turbine stages.
- Combustion in a turbine stage means high speed and accelerating flow.
- A cavity is used to create a low speed zone for mixing and flame holding.





Challenges:

- Ignition and flame-holding of spray flame in high-speed, high-acceleration flow;
- Mass/thermal transport between channel flow and cavity flow.

Collaboration with Professors Sirignano and Liu

Test Section

30 cm arc; 5 cm entrance height; 10 cm entrance width; 1 cm exit height



Flame Images

Deep & shallow cavity used on both pressure and suction sides

Re = 40000

Re = 40000

Propane, deep cavity, outer curve



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Electrical Properties of a Jet Diffusion Flame

- Point-to-Plane configuration
- H₃O⁺ is the dominant ion
- Ion-driven wind created by ions drifting through neutral gas after charge separation



Chemiluminescence & Schlieren



Electrical Effects: Co-Flow Diffusion Flame



- Nascent ions in flames have ~100ppb number density.
- Role of chemi-ions in combustion not fully understood.
- Many of the observed effects have been attributed to an ion-driven wind.
- As an actuator, an electric field acting on a flame through an ion/neutral gas coupling can enhance stability and flammability limits.
- As a sensor, ion detection can indicate the presence of a flame or incipient quenching.

Methane/Nitrogen Coflow Flame – Negative Mesh



1.2kV/cm

1kV/cm

0.8kV/cm

0.4kV/cm

4kV/cm

1.3kV/cm

- Ion drift upward initially destabilizes the flame.
- Flame oscillations begin near current saturation.

E

0kV/cm



Higher flow rates produce a lower ion current when scaled to flow rate.

- Oscillations occur with a downward directed field.
- Soot complicates the phenomenon

Ethylene Coflow Flame







Electrical Properties of Flames



-4kV field in microgravity

No field in 1g

Schlieren in 1g and microgravity with a field shows common features between buoyant forcing and electrical forcing.

Simulations – E-Field Flames

- Goals
 - Better knowledge of the chemical structure of the flame
 - Simulate concentrations of species profiles from CH4/ air laminar diffusion flame
 - Relate chemiluminescence to ion production
- Approach
 - 491 reactions steps mechanism assembled which includes: radicals, ions and chemiluminescence species (64)
 - CHEMKIN PRO package software 2D simulation of a cylindrical shear flow

Computational study of ions and excited state species in a methane/air co-flow diffusion flame



Simulations: E-field Flames

- OpenFOAM simulations using GRI Mech 3.0 with a chemi-ion mechanism
- Coflowing, axisymmetric geometry
- Solving for chemistry, N-S with electrostatic coupling, and Poisson's equation



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Sooting Behavior of Biofuels

with Prof. Pham, CSULA





Laser Induced Incandescence LII in a smoke lamp







Significance of Fuel Hydrates

- Enormous stored deposits of natural gas
 - Permafrost
 - Deep oceans
 - Potentially dangerous resource
 - CO₂ sequestration opportunity
- Applications
 - Pipelines
 - Gas storage

from http://gtresearchnews.gatech.edu/newsrelease/HYDRATES.htm via M.R. Walsh et al. / Energy Economics 31 (2009) 815–823



Hydrate Deposits



Experimental Apparatus



Methane Hydrate Burning



Powder Hydrate



Experimental Results

Mass of Original Sample	Hydrate Fraction	Water mass in original sample	methane mass in original sample	Ice remaining	water remaining	water evaporated	water vapor/methane molar ratio
35.2	0.87	31	4.2	5.2	10.5	5.3	3.3
57.9	0.86	51.2	6.7	5.9	36.4	8.9	1.2
37.6	0.87	33.2	4.4	5	9.6	18.6	3.7

Experimental Study of Water-Laden Fuel Mixtures Burning in a Non-Premixed Counterflow Configuration

Objective: Characterize non-premixed water/methane/air flames in a counterflow burner; distinguish thermal from chemical effects, e.g., when water vapor is injected into a methane stream. **Applications:** Methane hydrate combustion, emulsified water/fuel combustion, and LNG pool fires on bodies of water.

1. Experimental Apparatus

Components: (1) co-flow in both upper and lower ducts (2) electronic flow meters, (3) vaporizer, (4) cooling coil, honeycombs, and (4) heating and temperature control system.

2. Opposed-Jet Burner Conditions:

Non-Premixed Fuel/Oxidizer:

Upper Duct: Air oxidizer; Lower Duct: Fuel stream and water vapor. Burner:

Separation Distance between nozzles ~ 1-2 cm Flow Rates: Methane 4.4 l/min, Oxidixer 4.7 l/min

3. Flame Temperature Measurements

Thermocouple: Type –S [Pt-Pt 10% Ph], 0.20 mm bead size, fixed on a xyz stage.

Thin Filament Pyrometry (TFP)





Calibration of Pyrometry

Medium: Emission of light from K-type thermocouple inside furnace of known temperature. Sensor: Nikon D70 CCD. Red-Green blocking filter reduces RG intensities to below saturation.

Results







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Controlling Liquid Jet Breakup

A piezoelectric applies disturbance in the form of surface waves to the liquid jet by vibrating.

Different vibrations create different droplets patterns.

Controlling piezo's vibration, the breakup is controlled.

with Prof. Jabbari



Internal Combustion Engine Concepts



Advantages of HCCI: Low NOx & soot emission (satisfying 2010 emission standard without aftertreatment); 15-20% improvement in fuel economy

Disadvantages of HCCI: High HC and CO emissions under misfire; narrow operating regime; difficulty in ignition timing control

HCCI Test Engines, Single Cylinder Displacements





Next Step -- Spark Assisted Compression Ignition (SACI) Engine

Previously...

- 25cc, 4-stroke, air-cooled, single cylinder engine.
- Large S/V ratio rapid heat loss.
- Difficulty in ignition timing control.

Now...

- 49cc, 4-stroke, liquid-cooled, single cylinder engine.
- 49cc: Smaller S/V ratio, thus less heat through the surface.
- Liquid-cooled: Better engine wall temperature control.
- From HCCI to SACI: Better control on the ignition timing. Goals:
- Finding the operation regime of small-scale SACI: intake temperature, engine wall temperature, equivalence ratio
- Simple model to estimate temperature of the gas in the engine.
- Determine physical processes controlling SACI operation

Spray Characterization for Small-Scale Engine EFI System

Background

- Spray is widely used in many areas including agriculture, fire suppression, and combustion.
- Fuel injected IC engine performance depends on the spray. Typically, electronic Fuel Injection (EFI) systems provide higher thermal efficiency and power output, better transient response and also more precise Air/Fuel Ratio (AFR) control compared with traditional carburetor fuel feeding system.







Calibration fluid used as fuel simulent

injector

Typical structure of fuel injector



MicroSquirt EFI system

High-Speed Movie of the Starting Stage of Spray



Injector: compatible with Honda NPS50 scooter engine

Spray & Image parameters:

Camera: Phantom V4.3Fuel pressure: 45psi

- •Exposure time: 2µs
- •Frame rate (fps): 8510
- •Resolution: 128*512



Viscoelastic Fluid Mechanics

laties sheet

field of view

Non-intrusive velocity measurements (PIV), in rectangle channel



PIV measurements for quantifying Taylor instabilities



	and the second sec

Confocal microscopy

Near-field electrospinning





Chaining of 5µm fluorescent particle in viscoelastic fluids

Patterning with nanometer resolution with the help of viscoelastic ink

Droplet Vaporization in Isotropic Turbulence



Vertical Wind Tunnel Schematic

OBJECTIVE: Comparison of Measured and DNS Calculated Vaporization Rates of Droplets in Isotropic Turbulence

Droplet Vaporization in Isotropic Turbulence



Hot Wire Anemometry Measurements



Turbulence Generator



Power spectral density of U velocity at different downstream stations

Droplet Vaporization in Isotropic Turbulence

- Heptane droplets are introduced via calibrated syringe pump and initial size based on time between droplet release
- Droplets diameters measured using a DSLR camera near the test section exit





Droplet as imaged near bottom of test section

Suspended Heptane droplet evaporation

5 4.5 4 Air off grid off Air on grid off thin screen Air on grid off thick screen 3.5 \rightarrow Air on grid off thin/thick Diameter²(mm²) Air on grid off thick/thin -Air on grid off 3 Air on grid on thin screen Air on grid on thick screen 2.5 Air on grid on thin/thick Air on grid on thic / thin Air on grid on 2 1.5 -30 -10 -40 -20 0 time(sec)

Time between drops



Variation caused by syringe pump removed with damping reservoir

Dynamic Response of Teeth to a Dental Laser



Modern high speed drills have been produced since the mid 20th century. They can run at over 400,000 rpm, but cause much vibration. Starting in the late part of the 20th century, lasers were developed to be used in dentistry. The scope of this study is to find if lasers cause less vibration than traditional high speed drills.

The vibrometer and drill are facing the same direction, down towards the crown of the tooth. As the tooth vibrates from the drilling, the vibrometer measures the vertical motion of the tooth.



- Pink velocity of the tooth caused by a high speed drill.
- Blue = velocity caused by an Er,Cr:YSGG Laser at 300 mJ per pulse.

High-speed drill causes the tooth to move at 20 times the speed of a tooth ablated by a 300 mJ pulse of laser.



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